

Ocean Acidification

& its role in a changing climate



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WHY OCEANS MATTER

CLIMATE REGULATION

70% Covering 70% of the Earth's surface, the ocean transports heat from the equator to the poles, regulating our climate and weather patterns.



ECONOMY

\$282 billion Amount the U.S. ocean economy produces in goods and services. Ocean-dependent businesses employ almost 3 million people.



TRANSPORTATION

76% Percent of all U.S. trade involving some form of marine transportation.



RECREATION

From fishing to boating to kayaking and whale watching, the ocean provides us with so many unique activities.



FOOD

The ocean provides much more than just seafood. Ingredients from the sea are found in surprising foods such as peanut butter and soymilk.



MEDICINE

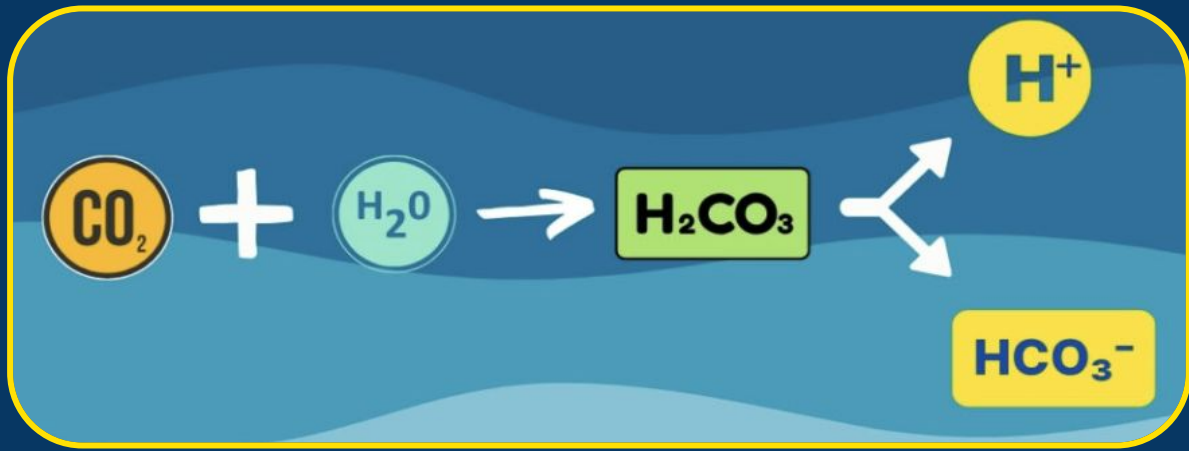
Many medicinal products come from the ocean, including ingredients that help fight cancer, arthritis, Alzheimer's disease, and heart disease.



CO₂

in the

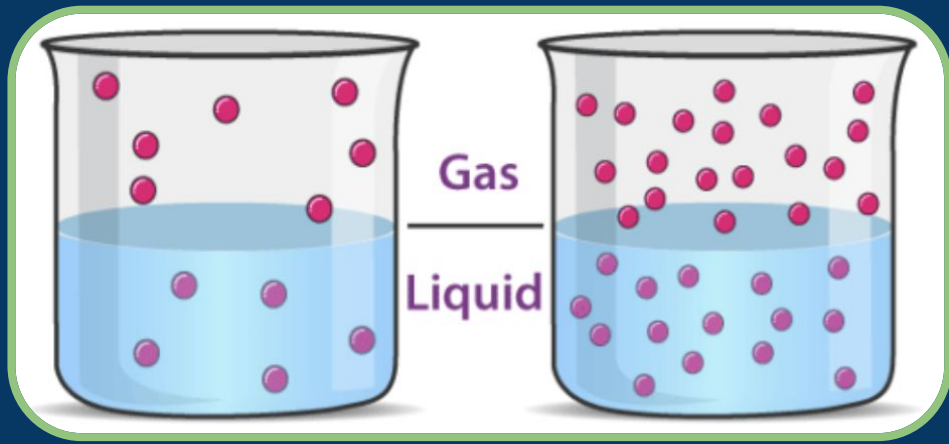
Sea



- Oceans can naturally solvate gaseous CO. Dissolved carbon dioxide + water react to produce carbonic acid (with an acidic hydrogen [pKa ~6.5], which is easily dissociable in water)
- Our oceans exist as a major 'carbon sink'. So, as industries and civilizations continue expanding, more CO₂ is emitted, thus more is absorbed by oceans.
- Currently, oceans have taken in ~30% of all human-emitted CO₂ ([ncei.gov](https://www.ncei.gov)), a massive amount!



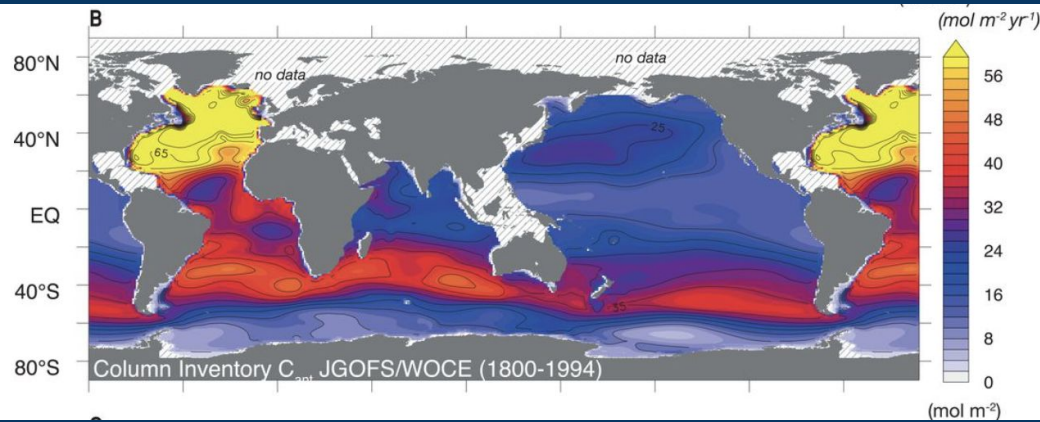
THE CHEMISTRY



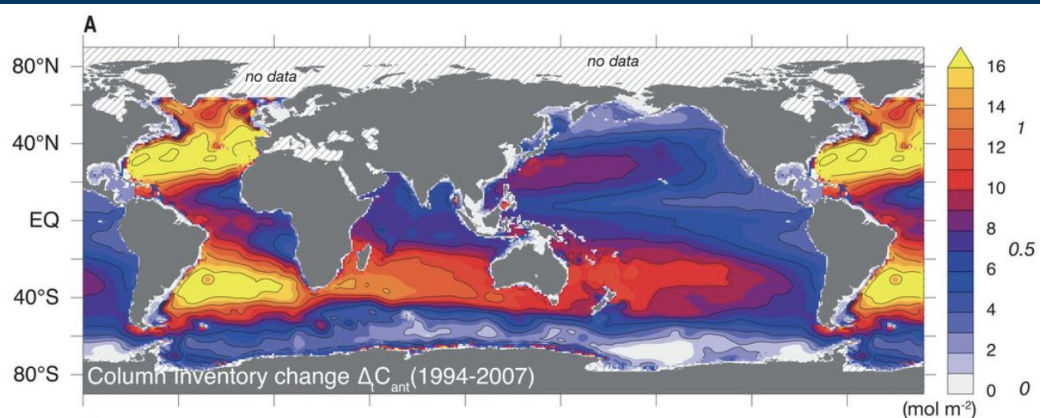
Henry's Law:

As the pressure of a gas over liquid increases, the amount of gas dissolved in the liquid increases proportionally

- So, as CO₂ increasingly populates the atmosphere, its fraction of the air pressure above our oceans increases. Thus, the amount of CO₂ dissolving into the oceans rises, and pH lowers.
- It is also important to understand that pH is a logarithmic scale. Ocean water has a natural, slightly alkaline pH (~8.1), which exists right around the acidity tipping point (7). So, small additions of acid can largely influence pH.
- Biological systems are designed to operate within a narrow range of pH, and acidity stresses these systems, threatening marine life and ecosystem wellbeing.



1800 to 1994



1994 to 2007

Extreme

ΔCO_2

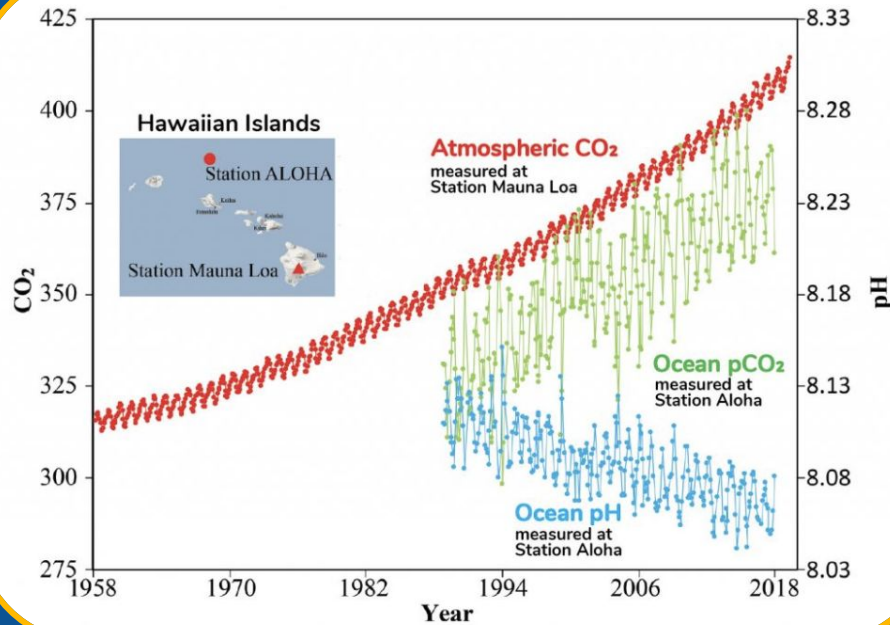
Mild

Locating **Extreme** Changes

Graphics depict measurement of ΔCO_2 Dissolved (anthropogenic) in oceans (0-3000m) ([science.org](https://www.science.org))

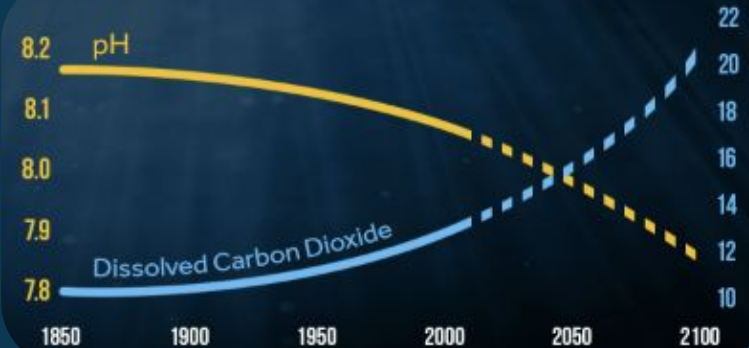
- Notice Non-uniform changes in CO₂. Certain areas show consistency in ΔCO_2 , as in the Northern Atlantic.
- Other areas emerge with extreme changes in recent years (Atlantic coast of South American and Southern coast of Africa).

Aqueous CO₂ & pH Over Time



-Data shows a strong upwards trend in dissolved gas, and decreasing of pH
-Notice significant fluctuations within the years.
Seasonal water conditions have a large impact on these values...

Study



Heating UP

An important nuance to Henry's Law:

As liquid temperature increases, solubility of gases will decrease. More energy (temperature) in a system means that dissolved gas molecules are more likely to pick up this energy, and escape back into the gas phase.

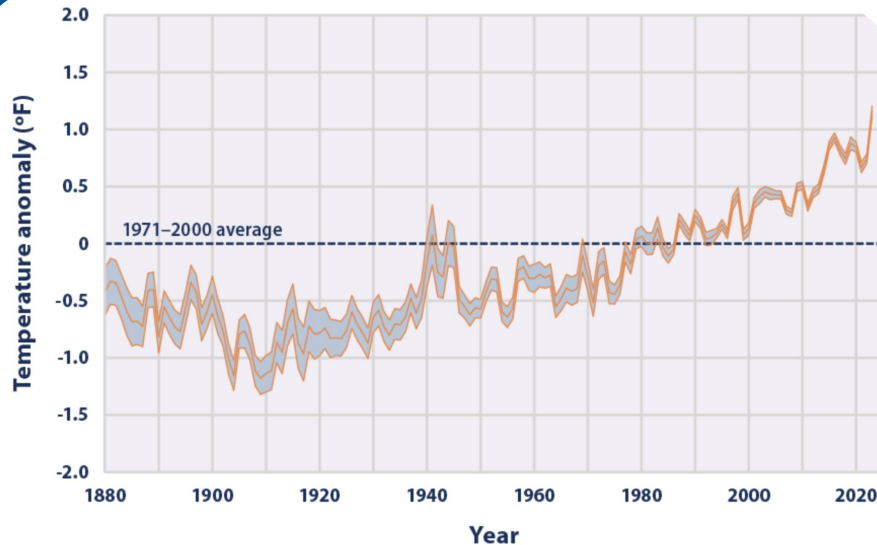


-You may think this is a positive factor, as rising ocean temps make it harder for acidification to propagate. However, it is important to consider that this is simply trading one problem for another. Rather than CO₂ acidifying the ocean, it will be released back into the atmosphere where it adds to GHGs, and drives the greenhouse effect.
-Also, consider that ~30% of all anthropogenic emissions has been solvated into the ocean, so that is a lot of potential CO₂ Gas

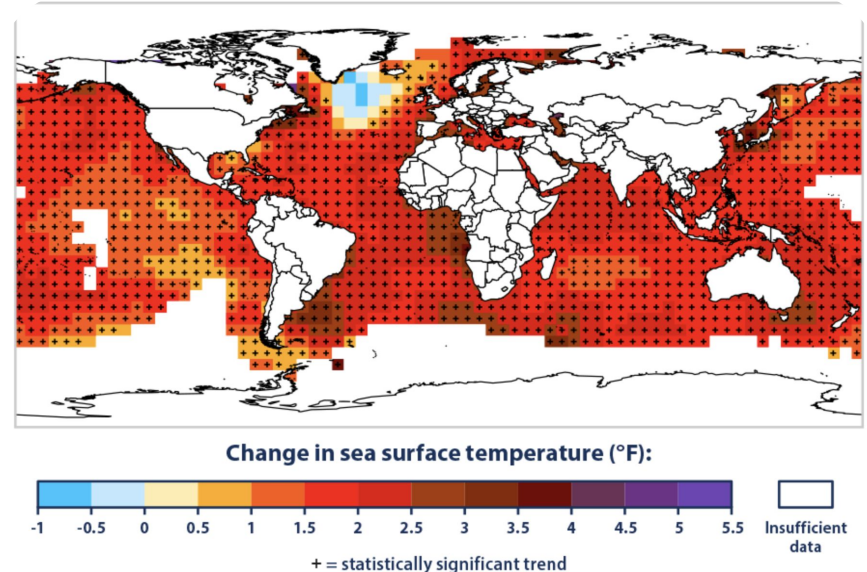
-By the same principle, warmer water also holds significantly less O₂. Unlike CO₂, dissolved oxygen is a fundamental resource for aquatic life. Less O₂ in water may translate to emerging hypoxia zones, where O₂ demand is greater than the supply, and ecosystem equilibrium is thrown off.
-Tropical regions, with preexisting warm waters are the areas most likely to experience immediate and more severe effects. This is a major risk factor, as much of the tropic reefs foster incredible biodiversity that is essential for the wellbeing of our oceans.

The Marine O_2 Squeeze

- In the past few decades, global dissolved oxygen levels have decrease by about 2% (4.8 petamoles, or 4.8×10^{15} moles)
- It is necessary to consider that this decrease is not evenly distributed. Some areas with more severe warming may lose 5%+ of oxygen, while cooler areas lose 0.5-1%.
- Such a significant change in conditions in certain locations strains marine habitats at both the local and basin scales.



Sea Surface Temperature



Ecological Impacts of Acidification

Coral Reefs and Shellfish

- Reduced calcium carbonate formation
- Severe acidity can dissolve existing coral structures
- 25% of all marine life depends on coral reefs for habitat and survival
- 500 million people rely on reef ecosystems for food and coastal protection from erosion

Food Web Disruption

- Acidification results in population decline of small animals like clams, oysters, and sea urchins
- Larger predators lose key food sources, creating shortages up the food chain and overall population decline
- Forces species to adapt, migrate to new areas, or face population collapse

Carbon Cycle Impact

As shells and coral bodies form less effectively, the ocean loses its ability to lock carbon away in these structures as a permanent sink, weakening the overall health and stability of marine ecosystems while increasing the amount of carbon in the atmosphere.

Algal Blooming

Nutrient Runoff

-When the level of nutrients such as nitrate and phosphate in our oceans are increased, they fuel large algal blooms



Algal Blooms

-Algal blooms increase respiration beneath the surface, creating hotspots where major O₂ consumption
At the end of blooms, large-scale decomposition produces more CO₂, which further decreases pH

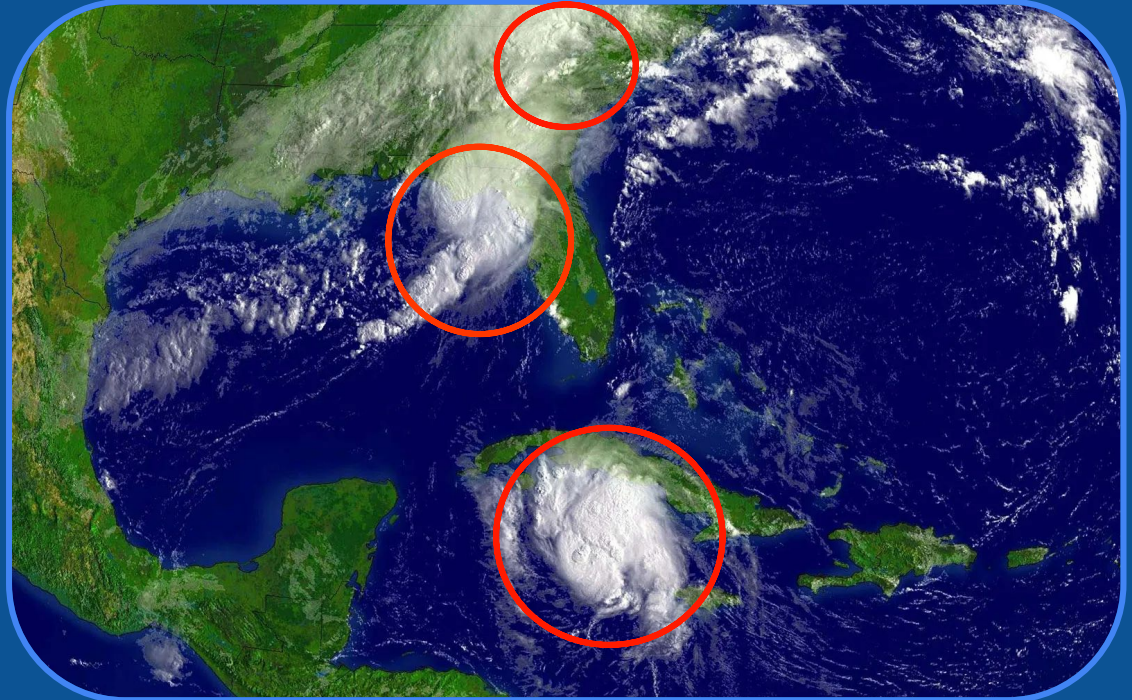
-Studies show that many harmful algal species bloom faster and produce more toxins in acidified waters, producing harmful effects



Atmospheric Effects:

-Warmer oceans give rise to greater quantities of water vapor in coastal areas. The atmosphere can hold ~7% more water vapor for every degree of warming.

-These humid regions propagate excessively heavy rainfall, and a much higher risk of frequent, intense tropical storms.



Human-related effects

- The shellfish industry increasing risk of collapse, ruining coastal communities globally, as well as a loss of hundreds of billions of dollars in the world economy

- Will increase risks in safety regarding consumption of anything from the sea, leading to higher health risks and premature deaths worldwide



Large Scale Solutions

OAE: Ocean alkalinity enhancement boosts the oceans ability to remove carbon from the air by increasing alkalinity.

Policy and The Globe:

- Intense international collaboration is a major required steps
- Economic relief to countries affected/unable to emit less emissions

Small SCALE Solutions



Seagrass restoration:

- Helps remove CO₂ from water
- Strengthens local ecosystems
- Provides shelter to endangered species

Modification of daily life:

- Utilization of carbon friendly transportation
- Joining local conservation efforts to preserve integrity of local coastlines

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